■塩酸の塩化水素ガス分圧

※出典：ソーダハンドブック
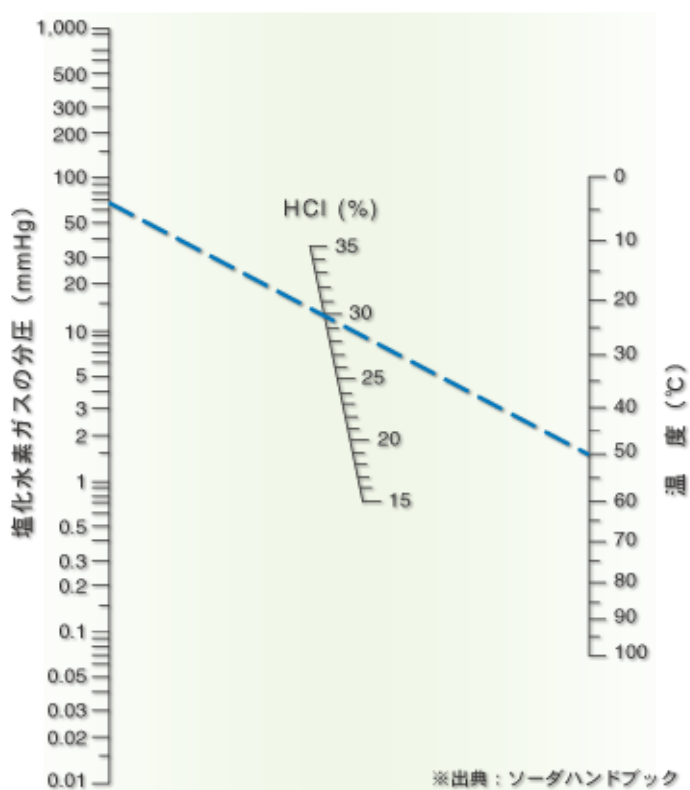
(注) NOMOGRAPH FOR PARTIAL PRESSURES OF HYDROGEN CHLORIDE
   IN AQUEOUS SOLUTIONS
   permits rapid solution of the Zeisberg equation
   $$\log P = 10.9525 - 0.009725\,W - 0.000868\,W^2 - \frac{4,557 - 65.3\,W}{t + 273}$$
   where P = partial pressure of HCl, mm. mercury
   W = weight per cent of HCl in aqueous solution
   t = temperature, °C